Proton Exchange Membrane Fuel Cell System Identification and Control –

Part I: System Dynamics, Modeling, Identification and Adaptive Control

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Outline

- Background
- Objective
- Fuel Cell System Modeling
- System Identification
- System Model with Time-Varying Parameters
- Adaptive Controller Design
- Summary and Conclusions

Background

Modeling system dynamics

- Steady state property:
 - the performance prediction for the purpose of designing cell components, choosing operating points and describing the steady state property.
- Transient dynamics:
 - air compressor, manifold filling dynamics, gas flow in the anode and cathode and its electro-chemical reactions.

<u>Complexity and difficulty: Modeling system dynamics</u> precisely is impossible

 electrochemistry, fluid dynamics, thermodynamics and heat transfer, time-varying and spatial physical properties, etc.

Complexity and Difficulty of Fuel Cell Control

Control of fuel cells

- Control of operation process
 - Valve, switch, fan, gas and water purge, over-voltage, over-current, emergency shut-off, etc.
- Control of output power toward the load
 electronics design of inverter and converter
 - Stable power supply
- <u>Control of fuel cell stack performance</u>
 - Less mentioned, no feasible model for controller design
 - Critical issue: stable power supply as well as efficient usage of fuel

Objective

To identify PEM fuel cell stack dynamics in a linearized, discrete-time, input and output model

To control the system through on-line parameter estimation and adaptive control

To adjust air and hydrogen flow rate to stablize the output voltage under various load requirements.

Fuel Cell System Fundamentals





Typical VI curve (steady-state)



Typical cell voltage vs. current density plots for PEM fuel cells and a common interpretation for the voltage drop.

Background of PEMFC System Modeling



Cathode Diffusion Dynamics

Continuity equation

where

Stefan Maxwell equation

$$\frac{\varepsilon_g}{RT}\frac{\partial p_i}{\partial t} + \frac{\partial N_i}{\partial y} = 0$$

$$\frac{\varepsilon_g}{\tau^2} \frac{\partial p_i}{\partial y} = \sum_{k=1}^3 \frac{RT}{p_c D_{ik}} (p_i N_k - p_k N_i)$$

Simplified as <u>cathode diffusion equation</u>

$$\frac{\partial p_1}{\partial t} = \omega \frac{\partial^2 p_1}{\partial \xi^2} - \psi \frac{j_r}{4F} \frac{\partial p_1}{\partial \xi}$$

$$\omega = \frac{1}{\tau^2 L_d^2 \left((p_{sat} / d_{12}) + (p_c - p_{sat}) / d_{13} \right)} \quad \psi = \frac{RT}{\varepsilon_g L_d \left(p_c - p_{sat} \right)}$$

pde: variation of partial pressures in terms of space and time, flow rate, material and geometric property of diffusion layers

Cathode Kinetics

Butler-Volmer equation

$$j_r = j_0 A_r \left\{ \frac{p_1}{p_{10}} \begin{bmatrix} H^+ \end{bmatrix}_0 exp\left(\frac{\eta}{b}\right) - 1 \right\}$$

• Overpotential on cathode $\eta = E_{\theta C} - \Delta \phi_{ce} = E_{\theta} - V_{cell} - R_{ohm} j$

Current density charge on double-layer

$$j = j_r + C_{dl} \frac{\partial \eta}{\partial t}$$

pde: relationship between current density, over-potential, proton concentration, catalyst contact area, etc.

Proton concentration dynamics

where
$$\begin{aligned} u \left(-\frac{\partial c_{H^+}}{\partial t} \right) \cdot \frac{\partial c_{H^+}}{\partial t} + \frac{c_{H^+}}{\tau_{H^+}} &= \frac{1 + \alpha_{H^+} \cdot j^3}{\tau_{H^+}} \\ c_{H^+} &= \frac{[H^+]/[H^+]_0}{\tau_{H^+}} \end{aligned}$$

Nonlinear pde: j^3 , u(.) - Heaviside function

Nernst equation

$$E = E_{ref} + \frac{dE^{0}}{dT} (T - T_{ref}) + k \frac{RT}{2F} ln(P_{H_2} P_{O_2}^{\frac{1}{2}})$$

Where E^{0} is the open loop voltage, ...

Internal resistance

$$R_{ohm} = R_{ref} + \alpha_T (T - T_{ref})$$

Air compressor

$$J_{cp} \frac{d \omega_{cp}}{dt} = \tau_{motor} - \tau_{cp}$$

$$N_A = F(\omega_{cp})$$

Other dynamics

Mass transportation, energy conservation of the reactant flows and pressures in the cathode and anode, water condensation, evaporation and generation, as well as quasi-steady-state temperature profile.

In summary

PEMFC dynamic model features:

- Complex physical phenomenon: principles of electrochemistry, fluid dynamics, thermodynamics and heat transfer, etc.
- Nonlinear dynamics: partial differential equations in terms of space and time, material coefficients and universal constants
- Approximation under various assumptions and constraints.
- Time-varying parameters
- Multi-input multi-output (MIMO) system
- Subject to external disturbances and unmodeled dynamics

Linearized MIMO system

Air pump

Vcp

Flow rate of air N_A

(partial pressure of oxygen)

(air pressure)

Flow rate of hydrogen *N_H*



Fuel cell output current I_c

Internal resistance *R*

Fuel cell output voltage Vc

 $I_C = G_1 N_A + G_3 N_H$ $V_C = G_2 N_A + G_4 N_H + R \cdot I_C$

Linearized MIMO System (without considering the auxiliary input part)

$$G_1(s) = \frac{I_C}{N_A}$$

$$G_2(s) = \frac{V_C}{N_A}$$

$$G_3(s) = \frac{I_C}{N_H}$$

$$G_4(s) = \frac{V_C}{N_H}$$

MIMO Linear Time-varying Coupled Transfer functions

At some operating point

System identification

 Persistently exciting the system with pseudo-random binary sequence (PRBS)



Output current



ARMAX Model (discrete time model of G_1 , G_2 , G_3 , and G_4)

$$A(q,k)y(k) = B(q,k)u(k) + C(q,k)W(k)$$

Auto-regression part: A(q,y) y(k)

$$A(q,k) = a_0(k) + a_1(k)q^{-1} + a_2(k)q^{-2} + \dots + a_r(k)q^{-r}$$

Moving average part: B(q,k) u(k)

$$B(q, k) = b_{1}(k)q^{-1} + b_{2}(k)q^{-2} + \cdots + b_{r}(k)q^{-r}$$

Auxiliary input part: C(q,k) e(k) –unmodelled dynamics

$$C(q,k) = C_{0}(k) + C_{1}(k)q^{-1} + C_{2}(k)q^{-2} + \cdots + C_{r}(k)q^{-r}$$

Recursive Least Squares Algorithm

Parameterization $\hat{y}(k) = \phi^{T}(k-1)\hat{\theta}(k-1)$ $\phi^{T}(k-1) = [-y(k-1)\dots-y(k-m)u(k-1)\dots u(k-n)w(k-1)\dots w(k-r)]$ $\hat{\theta}^{T} = [\hat{a}_{1}\dots\hat{a}_{r}\hat{b}_{1}\dots\hat{b}_{r}\hat{c}_{1}\dots\hat{c}_{r}]$ $w(k) = y(k) - \hat{y}(k)$

Parameter estimation $\hat{\theta}(k) = \hat{\theta}(k-1) + K(k)[y(k) - \phi^{T}(k-1)\hat{\theta}(k-1)]$ $K(k) = P(k-1)\phi(k-1)[\lambda + \phi^{T}(k-1)P(k-1)\phi(k-1)]^{-1}$ $P(k) = [I - K(k)\phi^{T}(k-1)]P(k-1)/\lambda$

where λ is the forgetting factor

Experiments: Single Cell Specifications

- Rated power
- Rated voltage
- Rated current
- Active area per cell
- Anode reactant
- Cathode reactant
- Operating temperature
- Ambient temperature
- Current density
- Cell voltage

6W 0.6V 10 A 50 cm² (5 cm x 10 cm) Pure compressed hydrogen Humidified air (50C, 1 atm) 295K (22C) 293K (20C) 0.2A/cm² 0.63V

Determination of System Order



Hydrogen flow rate vs. fuel cell voltage

2nd order is assumed

Identified Frequency Response



 G_1 (I_c / N_A) Bandwidth: 0.005 rad/sec, Operation conditions: N_{H2} =0.5 SLPM, V_c =0.6V $G_2 (V_c / N_A)$ Bandwidth: 6.3e-4 rad/sec, Operation conditions: N_{H2} =0.5 SLPM, I_c =10A

Identified Frequency Response



 G_3 (I_c / N_H) Bandwidth: 0.814 rad/sec, Operation conditions: $N_{air}=3$ SLPM, $V_c=0.6V$ $G_4 (V_c / N_H)$ Bandwidth: 0.011 rad/sec, Operation conditions: $N_{air}=3$ SLPM, $I_c=10A$

Response Curve Fitting with Fixed Parameters



Simulated response with transfer function of fixed parameters does not fit well with real or experimental response.

System Identification with Higher Orders (G₄)



2nd order approximation is reliable

Time varying characteristic



A set of constant parameters may not accurately reflect system inherently nonlinear and time-varying properties.

Response Curve Fitting with Time-Varying Parameters (G₁)



Response Curve Fitting with Time-Varying Parameters (G₂)



Response Curve Fitting with Time-Varying Parameters (G₃)



Response Curve Fitting with Time-Varying Parameters (G₄)



Response Curve Fitting with Time-Varying Parameters



This evidence of excellent response curve fitting with time-varying parameters provides a strong basis on the application of adaptive control to improve fuel cell system performance.

Adaptive control strategy



Adaptive Controller Design

Fuel cell 2nd order model

Controller model

$$G(z) = \frac{b_0 z + b_1}{z^2 + a_1 z + a_2}$$

$$G_{c}(z) = \frac{B_{0} + B_{1}z}{A_{0} + A_{1}z}$$

$$T(z) = \frac{G(z)G_c(z)}{1 + G(z)G_c(z)}$$

Desired closed loop poles

Closed-loop transfer function

F(z) = (A0+A1z)(z2+a1z+a2)+(B0+B1z)(b0z+b1)= (z+a)(z+b+jc)(z+b-jc),

where *a*, *-b+jc*, *-b-jc* are desired closed-loop poles.

Adaptive Control Simulation Configuration (Simulink)



Simulation

$(G_2$ between cell voltage V_c and air flow rate N_A)



The plant varies at 150, 200, 250, 300, 350, 400, and 450 second.



(very exciting result)



Experiments

(Block diagram of experiment)



Experiments

(specifications)

- Fuel cell stack: 100W, 12V
- Fuel cell model:
 - 1st order <u>ARMA</u> model instead of ARMAX
- Adaptive controller: 1st order
- Electronic load-meter: output current is adjustable

Control objective:

To control the air flowrate to regulate the output cell voltage to a desired level by fixing the hydrogen flowrate while the load current varies

Controller with Time-Invariant Parameters



Adaptive Control Parameter Variation



oad setting |----1A----|--2A--|--3A---|--4A---|---5A--------



Adaptive Control Performance of fuel cell



load setting|----1A----|--2A--|--3A---|--4A---|---5A-------



Summary and Conclusions

- System identification with a linear time-varying ARMA model is successfully performed
- First order adaptive control is verified sufficient to regulate system output voltage to a desired level

Future task:

- MIMO adaptive control to adjust air flowrate and hydrogen flowrate to reach the desired voltage and load current
- Higher order model for system identification and adaptive controller design
- Control of cell temperature and gas humidity
- Application to the higher power fuel cell
- Implementation on electric vehicles

Thank you for your attention!

2nd Order System Approximation

- Linear dynamics around a certain operating point – for a short time span
- Sufficient information on transient time response overshoot, rise time, etc.
- Frequency domain information bandwidth, output/input magnitude ad phase
- Basis of controller design